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On the Stability of a Single-Turn α-Helix: The Single versus Multiconformation Problem

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Insights into the design-based folding of peptides and proteins carry the potential to contribute to the long-standing protein folding problem, the development of new materials, and the conception of novel drug entities. Efforts to conscript β -peptides for roles previously served by α -peptides¹ and to refold natural proteins into unnatural shapes,² for example, represent two fruitful and recently traversed discovery pathways. In the same spirit, mere fragments of well-known secondary structures are being coaxed to adopt three-dimensional shapes normally reserved for larger molecules. Peptides rich with side chains bearing a propensity for α -helix formation rarely do so when the molecule contains less than 15 residues. Nonetheless, strategies for stabilizing α -helical peptides have taken advantage of the intrinsic helix dipole, capping motifs, organic templates, hydrophobic interactions, salt bridges, metal ion chelation, unnatural amino acids, and covalent side chain tethers.^{3,4}

A particularly striking example of the latter is the recent report of the first stable single α -helical turn in solution. The work employed a palladium-centered clip to bridge histidines at the ends of the pentapeptide Ac-HAAAH-NH₂ to give [Pd(en)Ac-HAAAH-NH₂]²⁺ (1).⁴



Although the 22-membered ring includes four amide bonds, two imidazole rings, and the inherent capacity for three *i*, $i + 4 \alpha$ -helical hydrogen bonds, it still retains 14 easily rotated single bonds. Nonetheless, interpretation of the compound's geometry-discerning NMR properties (44 ROE distances and three ³*J*_{HH}'s), assisted by H-bond constrained simulated annealing with XPLOR, led to the proposal of a well-defined α -helix with two orientations of the C-terminal amide.⁴ Our examination of the conformational profile for **1**, on the contrary, suggests a much more complex ensemble with little or no contribution from the ideal helical turn structure.

While the previous authors performed their modeling studies with the Ac-HAAAH-NH₂ pentapeptide alone, we chose to include the metal clip in the context of the AMBER* force field and the GBSA/ H₂O continuum solvent model.⁵ Parametrization of the Pd center⁶ in AMBER* involved a density functional optimization of the dicationic complex **2** (DFT: Becke3LYP/LANL2DZ,⁷ Figure 1). Examination of the torsional potential around the Pd–N bond for





Figure 1. The $[Pd(NH_2(CH_2)_2NH_2)Im_2]^{2+}$ complex optimized by DFT and reproduced by the AMBER*/GBSA/H₂O force field; AMBER* selected distances (Å) and angles (deg).



Figure 2. (a) Superposition of idealized α -helical **1-h** (grey) and the Kelso et al. α -helical Ac-HAAAH-NH₂ (red) derived by simulated annealing.⁴ (b) Overlap of **1-h** (grey) and unconstrained optimized **1-h'** (blue); hydrogen bonds in yellow. H-bond **i** is *i*, *i* + 4; **ii** *i*, *i* + 3.

a single imidazole with the same method provided a rotation barrier of 3.0 or 3.7 kcal/mol depending on the in-plane orientation of the rotating heterocyclic ligand. Both geometric and energetic features were incorporated into AMBER* as illustrated by its reproduction of the global minimum DFT structure (Figure 1).

Structure 2 was augmented with Ac-HAAAH-NH₂ (3) to give the linkage isomer of [Pd(en)(peptide)]²⁺ that is purported to form an α -helix in water (1),⁴ and then subjected to a 75 000 step Monte Carlo conformational search7 with AMBER*/GBSA/H2O and a 10 kcal/mol energy cutoff. The resulting 8850 optimized conformers were supplemented with two helical conformations. The first was generated by constrained torsional optimization of 1 to give the idealized α -helical form with two *i*, *i* + 4 hydrogen bonds, **1-h**. The metal complex is superimposed on the α -helical peptide in Figure 2a. The second related conformer was obtained by unconstrained optimization of 1-h providing 1-h', 6.0 kcal/mol lower in energy. It retains aspects of the helical features, but a somewhat different hydrogen-bonding pattern (one each i, i + 3 and i, i + 4H-bonds, Figure 2b, blue). The combined 8852 conformers and the NMR parameters measured by Kelso et al.4 (44 ROE distances and five ${}^{3}J_{\rm HH}$'s) were subjected to a NAMFIS⁸ conformational deconvolution, resulting in an eight conformer "best fit" of the data



Figure 3. Inverse [γ]-turn conformer **a** is predicted to have a 55% population in solution. The hydrogen bond, indicated by a dashed yellow line, and the sequestered His5 NH (**i**) satisfy reported VT-NMR data.⁴

(SSD⁹ = 102) with populations ranging from 2 to 55%. The most populated conformer (**a**, 55%, Figure 3) possesses one *i*, *i* + 2 hydrogen bond, signifying the presence of an inverse γ -turn.^{7,10-13} The intramolecular hydrogen bond at Ala4 is consistent with the reported variable temperature NH shifts,⁴ as is the sequestered His5 NH directed into the molecular cavity (Figure 3). The next three most-populated conformers also sustain a γ -turn (**b**, **c**, **d**; 16, 7, and 4%, respectively). The four γ -turn conformers differ qualitatively by at least one torsional angle.⁷ The fifth conformer (**e**, 3%) is a β -turn in which His5 is the NH H-bond donor to the Ala2 C=O. While 1-**h** is not among the NAMFIS conformers, 1-**h'** appears as the sixth most populated form at 3%. Submission of the 1-**h** and 1-**h'** pair alone to a NAMFIS matching of the same NMR data yields only 1-**h'** (100%) with an SSD = 171.

To evaluate the situation in the context of constraint-guided simulated annealing, we presented the linear structure of peptide 3, the NMR constraints, and the assumption of two i, i + 4 hydrogen bonds to CNSsolve/XPLOR.14 The 15 resulting structures and the optimized average are α -helical similar to Figure 2a (red). The simulated annealing exercise was repeated with the same NMR variables, but with H-bonding assumptions corresponding to 1-h'. The desired structure was obtained directly. Within the context of CNSsolve, 1-h' satisfies the data and exhibits constraint violations similar to those of the ideal α -helical structure. The constraint violation comparison is noteworthy considering that 1-h' is a local AMBER* minimum and satisfies only a subset of the NMR data. Conformer **a** is also located by the CNSsolve treatment.¹⁵ In view of these observations, the absence of the α -helical form from the NAMFIS conformers, and the fact that 1-h is a 6 kcal/mol destabilized virtual conformation (AMBER*), a single-turn α -helix appears to be an unrealistic solution to the NMR-derived metrics. It is clear that relaxation of the requirement that 1 adopts a single α -helical conformation, as necessitated by the CNSsolve/XPLOR treatment (see Supporting Information),^{4,14} permits the NMR spectra to be interpreted by NAMFIS in terms of a rapidly equilibrating mixture of eight conformers with a variety of hydrogen-bonding patterns.

This example illustrates a general problem facing workers applying 2-D NMR to the structure determination of small molecules in solution.¹⁶ Intuitive analysis of structure leading to the conclusion that only a single form exists in solution is a self-fulfilling exercise. Simulated annealing and related techniques which combine the totality of the NMR constraints and any user-conceived assumptions in a search for a lone structure will certainly deliver a family of such structures.¹⁷ However, unlike soluble proteins

where a single overall conformation is the norm, small molecules with one or more single bonds in general experience conformational averaging. While it may be attractive to computationally constrain a compound to a single conformation in solution, we believe it advisable to clearly demonstrate the fact outside the limits of assumption. NAMFIS is but one of a number of methods available¹⁸ to test whether an alternative multiconformational interpretation fits the data equally well or better.

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Supporting Information Available: Computational results (PDF). This material is available free of charge via the Internet at http://pubs.acs.org.

References

- (a) Hill, D. J.; Mio, M. J.; Prince, R. B.; Hughes, T. S.; Moore, J. S. *Chem. Rev.* 2001, 101, 3893-4012. (b) Cubberley, M. S.; Iverson, B. L. *Curr. Opin. Chem. Biol.* 2001, 5, 650-653. (c) Chung, Y. J.; Huck, B. R.; Christianson, L. A.; Stanger, H. E.; Krauthauser, S.; Powell, D. R.; Gellman, S. H. J. Am. Chem. Soc. 2001, 123, 5851-585. (d) Daura, X.; Gademann, K.; Schafer, H.; Jaun, B.; Seebach, D.; van Gunsteren, W. F. J. Am. Chem. Soc. 2001, 123, 2393-2404.
- (2) Dalal, S.; Balasubramanian, S.; Regan, L. Nat. Struct. Biol. 1997, 4, 548– 552. Predki, P. F.; Regan, L. Biochemistry 1995, 34, 9834–9839.
- (3) Fairlie, D.; West, M.; Wong, A. Curr. Med. Chem. 1998, 5, 29-62.
- (4) Kelso, M. J.; Hoang, H. N.; Appleton, T. G.; Fairlie, D. P. J. Am. Chem. Soc. 2000, 122, 10488–10489 and references therein.
- (5) MacroModel 6.5; cf. http://www.schrodinger.com.
- (6) Pd has been implemented in the MM2 force field: (a) Norrby, P.-O.; Åkermark, B.; Hæffner, F.; Hansson, S.; Blomberg, M. J. Am. Chem. Soc. 1993, 115, 4859–4867. (b) Yates, P. C. J. Mol. Struct. (THEOCHEM) 1994, 303, 55–64.
- (7) See the Supporting Information for details.
- (8) Cicero, D. O.; Barbato, G.; Bazzo, R. J. Am. Chem. Soc. 1995, 117, 1027-1033.
- (9) The sum of square differences gives the goodness of fit; cf. refs 8 and 17.
- (10) γ-Turns are about one-third as frequent as helices in proteins¹¹ and participate in [γ][γ]-turns,¹² while cyclic peptides commonly sustain γ-turns.¹³
- (11) cf. www.cdfd.org.in/dsmp.html.
- (12) (a) Guruprasad, K.; Prasad, M. S.; Kumar, G. R. J. Pept. Res. 2000, 56, 250–263. (b) Guruprasad, K.; Prasad, M. S.; Kumar, G. R. J. Pept. Res. 2001, 57, 292–300.
- (13) Wermuth, J.; Goodman, S. L.; Jonczyk, A.; Kessler, H. J. Am. Chem. Soc. 1997, 119, 1328–1335. Stachel, S. J.; Hu, H.; Van, Q. N.; Shaka, A. J.; Van Vranken, D. L. Bioorg. Med. Chem. 1998, 6, 1439–1446. Lindman, S.; Lindeberg, G.; Gogoll, A.; Nyberg, F.; Karlen, A.; Hallberg, A. Bioorg. Med. Chem. 2001, 9, 763–772. Prabhakaran, E. N.; Rajesh, V.; Dubey, S.; Iqbal, J. Tetrahedron Lett. 2001, 42, 339–342. Hedenstrom, M.; Yuan, Z.; Brickmann, K.; Carlsson, J.; Ekholm, K.; Johansson, B.; Kreutz, E.; Nilsson, A.; Sethson, I.; Kihlberg, J. J. Med. Chem. 2002, 45, 2501–2511.
- (14) CNSsolve (an XPLOR upgrade): Brunger, A. T.; Adams, P. D.; Clore, G. M.; Delano, W. L.; Gros, P.; Grosse-Kunstleve, R. W.; Jiang, J.-S.; Kuszewski, J.; Nilges, M.; Pannu, N. S.; Read, R. J.; Rice, L. M.; Simonson, T.; Warren G. L., http://cns.csb.yale.edu/v1.0/.
- (15) Conformers closely related to γ-turn a were identified by CNSsolve when three of the 44 NOEs were excluded; see Supporting Information for the rationale.
- (16) (a) Snyder, J. P.; Nevins, N.; Cicero, D. O.; Jansen, J. J. Am. Chem. Soc. 2000, 122, 724–725. (b) Monteagudo, E.; Cicero, D. O.; Cornett, B.; Myles, D. C.; Snyder, J. P. J. Am. Chem. Soc. 2001, 123, 6929–6930.
- (17) Nevins, N.; Cicero, D.; Snyder, J. P. J. Org. Chem. 1999, 64, 3979-3986.
- (18) (a) Landis, C.; Allured, V. S. J. Am. Chem. Soc. 1991, 113, 9493-9499. Landis, C. R.; Luck, L. L.; Wright, J. M. J. Magn. Reson., Ser. B 1995, 109, 44-59. Wright, J. M.; Landis, C. R.; Ros M. A. M. P.; Horton, A. D. Organometallics 1998, 17, 5031-5040. (b) Nikiforovich, G. V.; Vesterman, B. G.; Betins, J. Biophys. Chem. 1988, 31, 101-106. (c) Nikiforovich, G. V.; Kover, K. E.; Zhang, W.-J.; Marshall, G. R. J. Am. Chem. Soc. 2000, 122, 3262-3273. (d) Mierke, D. F.; Kurz, M.; Kessler, H. J. Am. Chem. Soc. 1994, 116, 1042-1049. (e) Cuniase, P.; Raynal, I.; Yiotakis, A. J. Am. Chem. Soc. 1997, 119, 5239-5248.

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